# Methyl 3-(triphenylphosphoranylideneamino)pyrazine-2carboxylate: synthesis, crystal structure and use in pteridin-4(3H)ones synthesis ${ }^{1}$ 

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The title iminophosphorane 6 has been prepared from methyl 3-aminopyrazine-2-carboxylate $\mathbf{1}$ by a modified Kirsanov method using a triphenylphosphine-hexachloroethane-triethylamine system. The crystal structure of 6 illustrated an iminophosphorane function ( $\mathrm{N}=\mathrm{P}$ bond). 2,3-Disubstituted pteridin$4(3 H)$-ones were obtained in a one-pot reaction of 6 with isocyanates, followed by heterocyclization on addition of alcohols or amines.

Over the past decade, the aza-Wittig reactions of aza-ylides have received increased attention in view of their utility in the synthesis of $\mathrm{C}=\mathrm{N}$ (imine) bond-containing compounds, in particular nitrogen heterocyclic compounds. ${ }^{2,3}$ The key intermediate iminophosphorane can be conveniently generated by a Staudinger reaction from azide derivatives and phosphorus(III) reagents such as triphenylphosphine or by a Kirsanov reaction from primary amines and phosphorus(v) reagents such as dichlorotriphenylphosphorane. ${ }^{4}$ Iminophosphoranes react with carbonyl compounds to form imines, and with isocyanates, isothiocyanates, carbon dioxide and carbon disulfide, to afford the corresponding heterocumulenes. ${ }^{5}$ In addition, the reactivity of iminophosphoranes is widely variable depending on the N and P substituents as well as the carbonyl function. We have demonstrated recently that the intramolecular aza-Wittig reaction is a powerful tool for the synthesis of $5-7$ membered nitrogen heterocycles such as oxazoles, ${ }^{6}$ imidazolinones, ${ }^{7}$ iminolactams, ${ }^{8}$ quinazolin- $4(3 H)$-ones, ${ }^{7,9-12}$ and 1,4-benzodiazepin- 5 -ones. ${ }^{13}$ On the other hand, an intermolecular aza-Wittig reaction followed by electrocyclization, intramolecular cycloaddition or heterocyclization, i.e. the tandem aza-Wittig and cyclization sequence, has been utilized for the synthesis of pyridines, pyrimidines and many other important nitrogen heterocycles by Molina, ${ }^{2 b}$ Wamhoff, ${ }^{14}$ and Motoki et al. ${ }^{15}$ Also, $N$-vinyliminophosphoranes have been utilized for the synthesis of heterocycles by Nitta et al. ${ }^{16}$ We became interested in the preparation of $N$-heteroaryliminophosphoranes because these species have been little studied, ${ }^{4}$ and are promising building blocks for the synthesis of nitrogen heterocycles. Fused heterocycles have been prepared via such iminophosphorane intermediates recently ${ }^{2,14}$ and we have reported the synthesis of pteridin- $4(3 \mathrm{H})$-ones by such a route in a preliminary communication. ${ }^{17}$ This paper reports detailed results on the synthesis of methyl 3-(triphenylphosphoranyl-ideneamino)pyrazine-2-carboxylate 6 , together with a crystal structure of this compound and its use in the synthesis of 2,3disubstituted pteridin-4(3H)-ones. Pteridine derivatives are of importance since their tetrahydro derivatives (e.g. tetrahydrofolic acid) are coenzymes for neurotransmitters (catecholamines and indoleamines), ${ }^{18}$ pyrimidine nucleotides, ${ }^{19}$ and methionine; ${ }^{20}$ further, methotrexate and its analogues have been shown to possess potent, broad spectrum antitumour activity. ${ }^{21}$ Although there are several syntheses of the pteridine skeleton by condensation of guanidines, the synthesis of 2,3-disubstituted pteridin-4(3H)-ones by such routes is prob-
lematic. ${ }^{22}$ We report here a convenient synthesis of such compounds by utilizing an aza-Wittig reaction/heterocyclization sequence.

## Results and discussion

Direct azidation of 3-aminopyrazine-2-carboxylic acid by the standard method using sodium nitrite and sodium azide in aqueous acid failed to afford the corresponding azide, presumably because either the diazo intermediate decomposes and/or the products are difficult to extract. Attempted azidation of the methyl ester derivative, $\mathbf{1}$ under the similar conditions afforded only the side products $\mathbf{3}$ and $\mathbf{4}$ in moderate yields ( 49 and $40 \%$, respectively). These results indicated that the corresponding diazonium intermediate was too reactive and reacted with chloride ion and hydroxide ion in the solution before reacting with azide anion to produce azide 2 . As a second approach, the 3 -chloro derivative 3 , readily obtained from the 3-hydroxy derivative 4, was treated with sodium azide in DMF to give, by nucleophilic azidation, ${ }^{23}$ the tetrazolo $[1,5-a]$ pyrazine derivative 5 ( $68 \%$ ) (Scheme 1).

Recent deoxidative azidation of pyrazine $N$-oxide with trimethylsilyl azide is known to be ineffective for derivatives containing an electron-withdrawing substituent, e.g. a $\mathrm{CO}_{2} \mathrm{Me}$ group. ${ }^{23}$ Since azide-tetrazole equilibration is well documented, ${ }^{24}$ compounds 5 were treated with triphenylphosphine ( 1.1 equiv.) in benzene at reflux for 2 h to afford quantitatively the corresponding iminophosphorane 6, via the Staudinger reaction (Scheme 1). ${ }^{4}$ The iminophosphorane 6 was also obtained, and more conveniently in $96 \%$ yield, as pale yellow crystals by a modified Kirsanov reaction ${ }^{4}$ of methyl 3 -amino-pyrazine-2-carboxylate 1 with in situ generated dichlorotriphenylphosphorane using a hexachloroethane-triphenylphos-phine-triethylamine reagent system (Scheme 1). ${ }^{25}$ The same reaction using carbon tetrachloride instead of hexachloroethane gave 6 only in modest yield ( $42 \%$ ) presumably as a result of a side reaction. The molecular structure was supported by the spectral data (IR, ${ }^{1} \mathrm{H}$ NHR, ${ }^{13} \mathrm{C}$ NMR and mass spectrum) and an X-ray crystallographic analysis. As summarized in Fig. 1 and Table 1, the imino(triphenyl)phosphorane structure was supported by the similarity of the $\mathrm{P}-\mathrm{N}$ bond length and $\mathrm{N}-\mathrm{P}-\mathrm{C}$ bond angle to those reported for $p$-bromophenylimino(triphenyl)phosphorane ( $\mathrm{P}-\mathrm{N}$ bond length: $1.567 \AA, \mathrm{P}-\mathrm{N}-\mathrm{C}$ bond angle: $\left.124.2^{\circ}\right)^{26}$ and $N$-[2-(triphenylphosphoranylideneamino) benzoyl]-L-valine ethyl ester ( $1.601 \AA, 124.0^{\circ}$ ). ${ }^{27}$


Scheme 1


Fig. 1 X-Ray structure of the pyrazin-2-yliminophosphorane 6 showing the atom labelling

Table 1 Bond lengths and bond angles of 6

| Bond length $(\AA)$ |  |  |  |
| :--- | :--- | :--- | :--- |
| Bond angle $\left({ }^{\circ}\right)$ |  |  |  |
| $\mathrm{P}(1)-\mathrm{N}(3)$ | 1.60 | $\mathrm{~N}(3)-\mathrm{P}(1)-\mathrm{C}(7)$ | 116.0 |
| $\mathrm{P}(1)-\mathrm{C}(7)$ | 1.83 | $\mathrm{~N}(3)-\mathrm{P}(1)-\mathrm{C}(13)$ | 104.1 |
| $\mathrm{P}(1)-\mathrm{C}(13)$ | 1.81 | $\mathrm{~N}(3)-\mathrm{P}(1)-\mathrm{C}(19)$ | 112.0 |
| $\mathrm{P}(1)-\mathrm{C}(19)$ | 1.80 | $\mathrm{P}(1)-\mathrm{N}(3)-\mathrm{C}(2)$ | 125.0 |
| $\mathrm{C}(2)-\mathrm{N}(3)$ | 1.32 |  |  |

Table $2 \mathrm{R}=\mathrm{Ph}$

| Entry | R'Y $^{\prime}$ | Compd. | Yield (\%) $^{a}$ |
| :--- | :--- | :--- | :--- |
| 1 | MeO | $\mathbf{1 0 a}$ | 70 |
| 2 | EtO | $\mathbf{1 0 b}$ | 45 |
| 3 | PrO | $\mathbf{1 0 c}$ | 44 |
| 4 | $\mathrm{Pr}^{\mathrm{i} O}$ | $\mathbf{1 0 d}$ | 36 |
| 5 | $\mathrm{Et}_{\mathbf{2}} \mathrm{N}$ | $\mathbf{1 0 e}$ | 92 |

${ }^{a}$ Isolated yields.

In confirming the iminophosphorane structure of 6, we examined its reaction with phenyl isocyanate. A mixture of 6 and phenyl isocyanate ( 3.5 equiv.) in dry benzene was stirred at room temperature. After disappearance of 6 (TLC monitored), the products were purified on a silica gel column chromato-

Table $3 \quad \mathrm{R}=$ Alkyl, $\mathrm{R}^{\prime} \mathrm{Y}=\mathrm{MeO}$

| Entry | R | Compd. | Yield (\%) $^{a}$ |
| :--- | :--- | :--- | :--- |
| 1 | $\operatorname{Pr}$ | $\mathbf{1 0 f}$ | 85 |
| 2 | $\operatorname{Pr}^{\mathrm{i}}$ | $\mathbf{1 0 g}$ | Trace $^{b}$ |
| 3 | Bn | $\mathbf{1 0 h}$ | 44 |

${ }^{a}$ Isolated yields. ${ }^{b}$ Compound $\mathbf{9 b}$ was obtained instead of $\mathbf{1 0 g}$. $\mathbf{9 b}$ was 3-isopropyl-2-isopropylaminopteridin-4(3H)-one.

Table $4 \quad \mathrm{R}=$ Aryl, $\mathrm{R}^{\prime} \mathrm{Y}=\mathrm{MeO}$

| Entry | R | Compd. | Yield (\%) $^{a}$ |
| :--- | :--- | :--- | :--- |
| 1 | $4-\mathrm{CH}_{3} \mathrm{C}_{6} \mathrm{H}_{4}$ | $\mathbf{1 0 j}$ | 55 |
| 2 | $3-\mathrm{CH}_{3} \mathrm{C}_{6} \mathrm{H}_{4}$ | $\mathbf{1 0 j}$ | 55 |
| 3 | $2-\mathrm{CH}_{3} \mathrm{C}_{6} \mathrm{H}_{4}$ | $\mathbf{1 0 k}$ | 29 |
| 4 | $4-\mathrm{CH}_{3} \mathrm{OC}_{6} \mathrm{H}_{4}$ | $\mathbf{1 0 1}$ | 55 |
| 5 | $3-\mathrm{CH}_{3} \mathrm{OC}_{6} \mathrm{H}_{4}$ | $\mathbf{1 0 m}$ | 81 |
| 6 | $2-\mathrm{CH}_{3} \mathrm{OC}_{6} \mathrm{H}_{\mathbf{4}}$ | $\mathbf{1 0 n}$ | 53 |
| 7 | $3-\mathrm{CF}_{3} \mathrm{C}_{6} \mathrm{H}_{4}$ | $\mathbf{1 0 0}$ | 48 |
| 8 | 4-ClC $6 \mathrm{H}_{4}$ | $\mathbf{1 0 p}$ | 77 |
| 9 | 1-Naphthyl | $\mathbf{1 0 q}$ | 30 |

${ }^{a}$ Isolated yields.
graphy to give the urea derivative $\mathbf{8}(34 \%)$ and the unwanted pteridinone $9 \mathrm{a}(31 \%)$ the identity of which was established on the basis of spectral results. Isolation of these products indicates, most reasonably, that the reaction involves formation of the corresponding carbodiimide 7, followed by addition of aniline which might be generated via decomposition of phenyl isocyanate by moisture (Scheme 2). Encouraged by the above findings, we treated 7 with the appropriate alcohols or secondary amines in order to prepare other pteridin-4(3H)ones (Scheme 2). Generally, the carbodiimide 7 was formed from 6 with isocyanate ( 3.5 equiv.) in benzene at room temperature (TLC monitored) after which an alcohol or an amine (excess) was added and the mixture heated for 3 h . Purification of the reaction mixture by silica gel column chromatography gave 2,3 -disubstituted pteridin-4(3H)-one derivatives 10 in $29-92 \%$ overall yields (see Tables 2-4). In these stepwise reactions, completion of cyclization was difficult to determine because moisture-induced decomposition of the intermediate $\mathbf{B}$ to $\mathbf{1}$ and carbamates or ureas occurred during TLC (Scheme 2). Therefore, low-yield entries might well be improved by longer periods of heating. Additionally, the intermolecular aza-Wittig reaction of 6 with alkyl isocyanates $\left(\mathrm{R}=\operatorname{Pr}, \operatorname{Pr}^{\mathrm{i}}, \mathrm{PhCH}_{2}\right.$ and 1-adamantyl, etc.) was very slow at room temperature although proceeding smoothly at $40-140^{\circ} \mathrm{C}$. In the case of 1-adamantyl isocyanates, however, severe steric hindrance prevented reaction even after 7 days at $140^{\circ} \mathrm{C}$. Also, with isopropyl isocyanate at $140^{\circ} \mathrm{C}$, large amounts of unwanted side-product were produced.

| Entry | R | $\mathrm{R}^{\prime}$ | Compd. | 12 Yield (\%) ${ }^{\text {a }}$ | 13 Yield (\%) |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | Ph | $\mathrm{Pr}^{\text {i }}$ | a | 82 | ND ${ }^{\text {b }}$ |
| 2 | Ph | Allyl | b | 45 | 39 |
| 3 | Ph | Prop-2-ynyl | c | 32 | ND |
| 4 | $4-\mathrm{ClC}_{6} \mathrm{H}_{4}$ | Allyl | d | 55 | 16 |
| 5 | 4- $\mathrm{ClC}_{6} \mathrm{H}_{4}$ | Prop-2-ynyl | e | 18 | ND |
| 6 | $4-\mathrm{CH}_{3} \mathrm{OC}_{6} \mathrm{H}_{4}$ | Allyl | f | 28 | 20 |

${ }^{a}$ Isolated yields. ${ }^{b}$ ND Not detected.


## Scheme 2

Table 6

| Entry | R | Compd. | Yield (\%) $^{a}$ |
| :--- | :--- | :--- | :--- |
| 1 | Ph | $\mathbf{1 5 a}$ | 99 |
| 2 | $4-\mathrm{ClC}_{6} \mathrm{H}_{4}$ | $\mathbf{1 5 b}$ | 100 |
| 3 | $4-\mathrm{CH}_{3} \mathrm{OC}_{6} \mathrm{H}_{\mathbf{4}}$ | $\mathbf{1 5 c}$ | 87 |

" Isolated yields.
Results for the cyclization of 7 with primary amines are summarized in Table 5 . Two isomeric pteridin- $4(3 H)$-ones 12 and 13 may be produced in the reaction of 7 with primary amines via a guanidine-type intermediate 11 (Scheme 3). In fact,


Scheme 3
the reaction with allylamine gave 12 and 13 but the reaction with isopropylamine afforded only 12 , compound 13 not being formed for steric reasons.

2-Allylaminopteridin-4(3H)-one derivatives 12 were further converted into imidazolo[1,2-a]pteridines 15 by treatment with iodine in high yield (Scheme 4, Table 6). Pteridines, biologically important polyazaheterocycles, may usefully be synthesized by the route described above. ${ }^{28}$

In summary, we have demonstrated that the $N$-heteroaryliminophosphorane, methyl 3-(triphenylphosphoranylidene-amino)pyrazine-2-carboxylate 6, can be obtained in high yield as a relatively stable crystalline compound by a modified Kirsanov reaction. The iminophosphorane structure was evidenced by an X-ray crystallographic analysis. The iminophosphorane 6 was a useful intermediate for the synthesis of


Scheme 4
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2,3-disubstituted pteridin-4(3H)-ones by an intermolecular azaWittig reaction with isocyanates followed by heterocyclization with alcohols or amines. The further application of azaWittig reaction methodology to the synthesis of various fusedpyrimidine derivatives is in progress in our laboratories.

## Experimental

## General

Thin layer chromatography (TLC) was performed on E. Merck Kieselgel $60 \mathrm{~F}_{254}$ pre-coated silica plates ( 0.15 mm layer thickness). Melting points were determined with a Yanagimoto micro-melting-point hot-stage apparatus and are uncorrected. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra were obtained with a Varian GEMINI-200 or 500 spectrometer at 200 or 500 and 50 or 125 MHz , respectively, for samples in $\mathrm{CDCl}_{3}$ or $\left[{ }^{2} \mathrm{H}_{6}\right]$-DMSO solution with $\mathrm{Me}_{4} \mathrm{Si}$ as internal standard. Chemical shifts are reported in ppm ( $\delta$ ). Coupling constants, $J_{\mathrm{H}}$ and $J_{\mathrm{C}}$ values, are given in Hz. IR spectra were recorded on a JASCO FT/IR 5300 spectrophotometer. Mass spectra (MS) and high-resolution mass spectra (HRMS) were recorded on a JEOL JMS-AX 505 HA (EI and/or CI, 70 eV ). Microanalyses were performed with a Perkin-Elmer 2400S elemental analyser. Flash chromatography was performed with a silica gel column (Fuji Davison BW-300) eluted with mixed solvents [hexane(H), ethyl acetate(A)].
Reagents and solvents. Benzene was stored over Na. Alcohols or amines were stored over $3 \AA$ molecular sieves. Isocyanates were dried over $\mathrm{CaH}_{2}$, distilled, and stored over $3 \AA$ molecular sieves. All reactions were carried out under nitrogen. Methyl 3-aminopyrazine-2-carboxylate 1 was purchased from Tokyo Kasei Co., Ltd. This reagent was used without further purification.

## Preparation of azide derivatives from amine derivatives

To an ice-cooled solution of anthranilic acid ( $5.001 \mathrm{~g}, 36.47$ mmol ) in concentrated hydrochloric acid ( $18 \mathrm{~mol} \mathrm{dm}^{-3} ; 36 \mathrm{~cm}^{3}$ ) and water $\left(36 \mathrm{~cm}^{3}\right)$ was added dropwise a solution of sodium nitrite ( $2.776 \mathrm{~g}, 40.23 \mathrm{mmol}, 1.1$ equiv.) in water ( $30 \mathrm{~cm}^{3}$ ) at a rate such that the temperature of the reaction mixture remained $<5^{\circ} \mathrm{C}$. After completion of the nitrite addition, the diazonium solution was filtered and added dropwise to a stirred solution of sodium azide ( $2.608 \mathrm{~g}, 40.12 \mathrm{mmol}, 1.1$ equiv.) and sodium acetate $(36.00 \mathrm{~g})$ in water $\left(60 \mathrm{~cm}^{3}\right)$. The solution was stirred for 15 min after which it was acidified with concentrated hydrochloric acid to give $o$-azidobenzoic acid as white needles ( $4.100 \mathrm{~g}, 25.13 \mathrm{mmol}, 69 \%$ ). However, similar azidation of methyl 3-aminopyrazine-2-carboxylate $1(505 \mathrm{mg}, 3.30 \mathrm{mmol})$ as a starting material failed to afford an azide derivative but, instead, the esters 3 and 4 together with recovered ester 1 (23.3 $\mathrm{mg}, 0.15 \mathrm{mmol}, 5 \%$ ) were obtained after silica gel column chromatography ( $\mathrm{H}: \mathrm{A} 3: 1, \mathrm{v} / \mathrm{v}$ ).
Methyl 3-chloropyrazine-2-carboxylate 3. White solid (277 $\mathrm{mg}, 1.61 \mathrm{mmol}, 49 \%$ ), mp $39-40{ }^{\circ} \mathrm{C}$ (Found: C, $42.0 ; \mathrm{H}, 2.8$; $\mathrm{N}, 16.1 . \mathrm{C}_{6} \mathrm{H}_{5} \mathrm{ClN}_{2} \mathrm{O}_{2}$ requires C, $41.76 ; \mathrm{H}, 2.92 ; \mathrm{N}, 16.23 \%$ ); $R_{\mathrm{F}} 0.51$ (H:A $1: 1$ ); $v_{\text {max }}\left(\right.$ neat $/ \mathrm{cm}^{-1}$ 2957, 1744, 1551, 1528, $1447,1383,1298,1152,1071,849$ and $756 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200\right.$ $\mathrm{MHz}) 8.61(1 \mathrm{H}, \mathrm{d}, J 2.4), 8.56(1 \mathrm{H}, \mathrm{d}, J 2.4)$ and $4.05(3 \mathrm{H}$, s); $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 164.1,148.2,146.1,144.5,142.2$ and 53.6; $m / z 172\left(\mathrm{M}^{+}, 19 \%\right), 149(16), 144$ (20), 143 (12), 142 (61), 141 (31), 116 (28), 115(18), 114 (100), 113 (51) and 86 (12).
Methyl 3-hydroxypyrazine-2-carboxylate 4. Pale yellow solid ( $205 \mathrm{mg}, 1.33 \mathrm{mmol}, 40 \%$ ), mp $155-156^{\circ} \mathrm{C}$ (Found: C, 46.9 ; $\mathrm{H}, 3.8$; $\mathrm{N}, 18.2 . \mathrm{C}_{6} \mathrm{H}_{6} \mathrm{~N}_{2} \mathrm{O}_{3}$ requires $\mathrm{C}, 46.76$; $\mathrm{H}, 3.92$; N , $18.18 \%) ; R_{\mathrm{F}} 0.11(\mathrm{H}: \mathrm{A} 1: 1) ; v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 2951,1717,1682$, $1665,1597,1458,1321,1200,1155,837$ and $806 ; \delta_{\mathrm{H}}\left(\left[{ }^{2} \mathrm{H}_{6}\right]\right.$ DMSO, 200 MHz ) $12.86(1 \mathrm{H}, \mathrm{br}), 7.72(1 \mathrm{H}, \mathrm{d}, J 3.6), 7.47(1$ $\mathrm{H}, \mathrm{d}, J 3.6$ ) and $3.81(3 \mathrm{H}, \mathrm{s}) ; \delta_{\mathrm{c}}\left(\left[{ }^{2} \mathrm{H}_{6}\right]\right.$-DMSO, 50 MHz$)$ 164.7, 154.4, 145.5, 133.1, 124.0 and $52.4 ; m / z 154\left(\mathrm{M}^{+}, 62 \%\right)$, 124 (51), 123 (11), 96 (100), 95 (31), 94 (27), 68 (53) and 67 (11).

Synthesis of methyl 3-chloropyrazine-2-carboxylate 3 from methyl 3-hydroxypyrazine-2-carboxylate 4
To a solution of the ester $4(41.0 \mathrm{mg}, 0.266 \mathrm{mmol})$ in dry toluene $\left(4 \mathrm{~cm}^{3}\right)$ was added thionyl chloride $\left(0.02 \mathrm{~cm}^{3}, 0.276 \mathrm{mmol}, 1.0\right.$ equiv.) and a catalytic amount of DMF (one drop) by syringe. After the mixture had been stirred at $80^{\circ} \mathrm{C}$ under nitrogen for 3 h , it was evaporated under reduced pressure, and the residue was purified by column chromatography (silica gel, $\mathrm{H}: \mathrm{A} 2: 1$, $\mathrm{v} / \mathrm{v}$ as eluent) to afford the title ester $\mathbf{3}(36.8 \mathrm{mg}, 0.213 \mathrm{mmol}$, $80 \%$ ).

## Synthesis of methyl tetrazolo[1,5-a]pyrazine-4-carboxylate 5

A solution of the ester $\mathbf{3}(116 \mathrm{mg}, 0.67 \mathrm{mmol})$ and sodium azide ( $88 \mathrm{mg}, 1.35 \mathrm{mmol}, 2.0$ equiv.) in DMF ( $4 \mathrm{~cm}^{3}$ ) was stirred at $120^{\circ} \mathrm{C}$ under nitrogen for 1 h . After completion of the reaction (TLC), the mixture was diluted with water ( $10 \mathrm{~cm}^{3}$ ) and extracted with chloroform ( $50 \mathrm{~cm}^{3} \times 3$ ). The combined extracts were dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under reduced pressure to afford the crude product which was purified on a silica gel column using $\mathrm{H}: \mathrm{A}(1: 1, \mathrm{v} / \mathrm{v})$ as an eluent to give the title ester as a yellow solid $5(82 \mathrm{mg}, 0.46 \mathrm{mmol}, 68 \%), \mathrm{mp} 136-$ $137{ }^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}$, 179.0447. $\mathrm{C}_{6} \mathrm{H}_{5} \mathrm{~N}_{5} \mathrm{O}_{2}$ requires $M$, 179.0443); $R_{\mathrm{F}} 0.30(\mathrm{H}: \mathrm{A}, 1: 2) ; v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 3094,1736$, $1474,1439,1354,1327,1254,1146,1103,1076,1001$ and 804; $\delta_{\mathrm{H}}\left(\left[^{2} \mathrm{H}_{6}\right]\right.$-DMSO, 200 MHz$) 9.69(1 \mathrm{H}, \mathrm{d}, J 4.6), 8.69(1 \mathrm{H}, \mathrm{d}, J$ $4.6)$ and $4.05(3 \mathrm{H}, \mathrm{s}) ; \delta_{\mathrm{C}}\left(\left[{ }^{2} \mathrm{H}_{6}\right]\right.$-DMSO, 50 MHz$) 161.8,143.9$, 141.5, 133.0, 123.2 and $53.5 ; m / z 179\left(\mathrm{M}^{+}, 7 \%\right), 151$ (57), 148 (56), 121 (22), 106 (24), 95 (13), $94(11), 93(91), 92(11), 80(100)$, 79 (74), 69 (11), 67 (24), 66 (14) and 65 (17).

Synthesis of $N$-heteroaryl iminophosphoranes; methyl 2-(triphenylphosphoranylidene)aminopyrazine-3-carboxylate 6 To a mixture of the ester $\mathbf{1}(271 \mathrm{mg}, 1.77 \mathrm{mmol})$, hexachloroethane ( $627 \mathrm{mg}, 2.65 \mathrm{mmol}, 1.5$ equiv.) and triphenylphosphine ( $696 \mathrm{mg}, 2.66 \mathrm{mmol}, 1.5$ equiv.) in dry benzene ( $20 \mathrm{~cm}^{3}$ ) was added dropwise triethylamine ( 436 mg , $4.30 \mathrm{mmol}, 2.4$ equiv.). The resultant mixture was heated at reflux for 5 h under nitrogen, after which it was cooled, filtered to remove the precipitate and evaporated under reduced pressure to afford a solid residue which was purified on a silica gel column using AcOEt-hexane ( $1: 1, \mathrm{v} / \mathrm{v}$ ) as an eluent to give the corresponding iminophosphorane $6(702 \mathrm{mg}, 1.70 \mathrm{mmol}$, $96 \%$ ).

## Alternative synthesis of $N$-heteroaryl iminophosphoranes; the ester 6 from the tetrazole 5

A mixture of the ester $5(78 \mathrm{mg}, 0.44 \mathrm{mmol})$ and triphenylphosphine ( $126 \mathrm{mg}, 0.48 \mathrm{mmol}, 1.1$ equiv.) in dry benzene ( $10 \mathrm{~cm}^{3}$ ) was heated at reflux for 2 h under nitrogen. The mixture was then evaporated under reduced pressure to afford a residue which was purified on a silica gel column using $\mathrm{H}: \mathrm{A}(1: 1, \mathrm{v} / \mathrm{v})$ as an eluent to give the corresponding iminophosphorane $6(180 \mathrm{mg}, 0.44 \mathrm{mmol}, 100 \%$ ) as a pale yellow solid mp 148-149 ${ }^{\circ} \mathrm{C}$ (Found: C, $69.8 ; \mathrm{H}, 4.8 ; \mathrm{N}, 10.2$. $\mathrm{C}_{24} \mathrm{H}_{20} \mathrm{~N}_{3} \mathrm{O}_{2}$ P requires: C, $69.73 ; \mathrm{H}, 4.88 ; \mathrm{N}, 10.16 \%$ ); $R_{\mathrm{F}} 0.47$ ( $\mathrm{H}: \mathrm{A} 1: 1$ ); $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 3061,2948,1740,1555,1505,1458$, 1433, 1277, 1132, 1115, 721 and 694; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right.$ ) 7.81-7.93 ( $7 \mathrm{H}, \mathrm{m}$ ), $7.75(1 \mathrm{H}, \mathrm{d}, J 2.4), 7.40-7.59(9 \mathrm{H}, \mathrm{m})$ and $4.03(3 \mathrm{H}, \mathrm{s}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 167.1,158.9(\mathrm{~d}, J 6.5), 144.7$, 137.8 (d, J 22.9), 133.6 (d, $J 10.0$ ), 132.3 (d, J 1.3), 131.5, 129.4, (d, $J 100.5$ ), $128.8(\mathrm{~d}, J 12.5)$ and $52.4 ; m / z 413\left(\mathrm{M}^{+}, 23 \%\right), 399$ (25), 398 (100), 354 (20), 352 (8), 277 (6), 262 (12), 261 (6), 201 (7), 183 (25) and 108 (10).

## Synthesis of the pteridin-4(3H)-ones 10a-q: general procedure

To a solution of the iminophosphorane $\mathbf{6}(103 \mathrm{mg}, 0.25 \mathrm{mmol})$ in dry benzene ( $10 \mathrm{~cm}^{3}$ ) was added dropwise an appropriate isocyanate (ca. 3.5 equiv.) with exclusion of moisture. After the mixture had been stirred at room temperature overnight, the iminophosphorane 6 had disappeared (TLC) and it was
therefore treated with an appropriate alcohol or amine (ca. 10 equiv.). The resultant solution was stirred at $85^{\circ} \mathrm{C}$ (benzene reflux) for 3 h after which it was evaporated under reduced pressure and the solid residue was purified on a silica gel column using $\mathrm{H}: \mathrm{A}(1: 1 \rightarrow 1: 2 \rightarrow$ only $\mathrm{A}, \mathrm{v} / \mathrm{v})$ as eluents to give the pteridin- $4(3 H)$-one derivatives 10 (e.g. 10a; $44.6 \mathrm{mg}, 0.175 \mathrm{~mol}$, $70 \%$ ). In the absence of an alcohol or amine under these conditions (vide supra), none of the desired pteridin- $4(3 \mathrm{H})$-one derivatives were obtained but instead, the urea derivatives 8 and the unwanted pteridin- $4(3 H)$-one derivative 9a were produced.
Methyl 2-(phenylcarbamoylamino)pyrazine-3-carboxylate 8. Pale yellow solid ( $18.4 \mathrm{mg}, 68 \%$ ), mp $162-163{ }^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}$, 272.0916. $\mathrm{C}_{13} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{O}_{3}$ requires $M$, 272.0909); $R_{\mathrm{F}} 0.13$ ( $\mathrm{H}: \mathrm{A}$ 1:1); $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 2924,1698,1613,1561,1476,1447,1310$, $1262,1115,1024$ and $814 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 11.17(1 \mathrm{H}$, br), $10.30(1 \mathrm{H}, \mathrm{br}), 8.45(1 \mathrm{H}, \mathrm{d}, J 2.3), 8.37(1 \mathrm{H}, \mathrm{d}, J 2.3), 7.56-7.62$ ( $2 \mathrm{H}, \mathrm{m}$ ), 7.32-7.41 ( $2 \mathrm{H}, \mathrm{m}$ ), 7.08-7.17 ( $1 \mathrm{H}, \mathrm{m}$ ) and $4.08(3 \mathrm{H}$, s); $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 166.3,151.5,150.7,144.6,138.2,136.6$, $129.4,127.9,124.4,120.8$ and $53.8 ; m / z 272\left(\mathrm{M}^{+}, 52 \%\right), 153$ (76), 148 (11), 123 (12), 120 (22), 119 (27), 95 (59), 94 (16), 93 (100), 91 (10) and 77 (11).

2-Anilino-3-phenylpteridin-4(3H)-one 9a. Pale yellow solid $(9.8 \mathrm{mg}, 31 \%), \mathrm{mp} 245-246^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}, 315.1108$. $\mathrm{C}_{18} \mathrm{H}_{13} \mathrm{~N}_{5} \mathrm{O}$ requires: $M, 315.1120$ ); $R_{\mathrm{F}} 0.16$ (H:A 1:2); $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 2996,1703,1605,1589,1561,1474,1451,1408$, 1269, 1026, 750 and 694; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.79(1 \mathrm{H}, \mathrm{d}, J$ 2.1), $8.59(1 \mathrm{H}, \mathrm{d}, J 2.1), 7.66-7.78(3 \mathrm{H}, \mathrm{m}), 7.45-7.56(4 \mathrm{H}, \mathrm{m})$, 7.30-7.39 ( $2 \mathrm{H}, \mathrm{m}$ ), 7.12-7.20 ( $1 \mathrm{H}, \mathrm{m}$ ) and $6.26(1 \mathrm{H}, \mathrm{s})$; $\delta_{\mathrm{H}}\left(\left[{ }^{2} \mathrm{H}_{6}\right]\right.$-DMSO, 200 MHz ) $8.75(1 \mathrm{H}, \mathrm{d}, J 2.1), 8.50(1 \mathrm{H}, \mathrm{d}, J$ $2.1), 8.01(1 \mathrm{H}, \mathrm{s}), 7.53-7.69(5 \mathrm{H}, \mathrm{m}), 7.43-7.49(2 \mathrm{H}, \mathrm{m}), 7.29-$ $7.34(2 \mathrm{H}, \mathrm{m})$ and 7.12-7.20 ( $1 \mathrm{H}, \mathrm{m})$; $\delta_{\mathrm{C}}\left(\left[{ }^{2} \mathrm{H}_{6}\right]\right.$-DMSO, 50 $\mathrm{MHz}) 161.4,155.7,151.4,150.4,140.6,138.3,134.7,130.5$, 130.1, 129.6, 128.5 and 125.2 (two quaternary carbons were not detected); $m / z 315\left(\mathrm{M}^{+}, 37 \%\right), 314$ (100), 286 (2), 238 (2), 223 (4), 195 (7), 169 (2), 168 (2), 92 (2) and 77 (14).

2-Methoxy-3-phenylpteridin-4(3H)-one 10a. Pale yellow solid $(44.6 \mathrm{mg}, 70 \%), \mathrm{mp} 193-195^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}, 254.0804$. $\mathrm{C}_{13} \mathrm{H}_{10} \mathrm{~N}_{4} \mathrm{O}_{2}$ requires $M, 254.0804$ ); $R_{\mathrm{F}} 0.21$ (A); $v_{\text {max }}(\mathrm{K}-$ $\mathrm{Br}) / \mathrm{cm}^{-1} 3061,1707,1605,1595,1561,1537,1447,1410,1354$, $1290,1090,961$ and $694 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.87(1 \mathrm{H}, \mathrm{d}, J$ 2.0 ), 8.73 ( $1 \mathrm{H}, \mathrm{d}, J 2.0$ ), $7.26-7.31(3 \mathrm{H}, \mathrm{m}), 7.15-7.17(2 \mathrm{H}, \mathrm{m})$ and $4.11(3 \mathrm{H}, \mathrm{s}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 161.5,156.1,154.6,150.4$, $143.1134 .1131 .9,130.0,129.9,128.3$ and $57.1 ; m / z 254\left(\mathrm{M}^{+}\right.$, $100 \%$ ), 253 (10), 239 (18), 237 (26), 233 (7), 196 (5), 195 (15), 134 (5) and 119 (21).

2-Ethoxy-3-phenylpteridin-4(3H)-one 10b. Pale yellow solid ( $15.6 \mathrm{mg}, 45 \%$ ), $\mathrm{mp} 205-208{ }^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}, 268.0961$. $\mathrm{C}_{14} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{O}_{2}$ requires $M, 268.0960$ ); $R_{\mathrm{F}} 0.25(\mathrm{~A}) ; v_{\text {max }}(\mathrm{K}-$ $\mathrm{Br}) / \mathrm{cm}^{-1} 2922,1715,1589,1561,1535,1426,1393,1329,1296$, $1090,1011,901$ and $698 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.86(1 \mathrm{H}, \mathrm{d}, J$ $2.0), 8.71(1 \mathrm{H}, \mathrm{d}, J 2.0), 7.52-7.56(3 \mathrm{H}, \mathrm{m}), 7.25-7.30(2 \mathrm{H}, \mathrm{m})$, $4.62(2 \mathrm{H}, \mathrm{q}, J 7.1)$ and $1.30(3 \mathrm{H}, \mathrm{t}, J 7.1) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right)$ $161.6,155.5,154.7,150.4,143.0,134.6131 .8,129.9,129.7,128.3$, 66.4 and $14.1 ; m / z 268\left(\mathrm{M}^{+}, 39 \%\right), 240(100), 239$ (15), 212 (13), 195 (12), 121 (15), 120 (26), 119 (14), 105 (14), 104 (34), 93 (16) and 77 (17).

3-Phenyl-2-propoxypteridin-4(3H)-one 10c. Pale yellow solid $(10.3 \mathrm{mg}, 44 \%), \mathrm{mp} \quad 185-187^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}, 282.1121$. $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{~N}_{4} \mathrm{O}_{2}$ requires $M, 282.1117$ ); $R_{\mathrm{F}} 0.25$ (A); $v_{\text {max }}(\mathrm{K}-$ $\mathrm{Br}) / \mathrm{cm}^{-1} 2924,1713,1593,1561,1539,1476,1426,1391,1331$, 1294, 1092, 937, 824 and 694; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.85(1 \mathrm{H}, \mathrm{d}$, $J 2.1), 8.71(1 \mathrm{H}, \mathrm{d}, J 2.1), 7.49-7.56(3 \mathrm{H}, \mathrm{m}), 7.26-7.31(2 \mathrm{H}$, $\mathrm{m}), 4.51(2 \mathrm{H}, \mathrm{t}, J 6.5), 1.67(2 \mathrm{H}, \mathrm{tq}, J 6.5,7.4)$ and $0.82(3 \mathrm{H}, \mathrm{t}, J$ 7.4); $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 161.6,155.8,154.7,150.3,142.9134 .6$ $131.8,129.9,129.7,128.2,71.9,21.8$ and $10.1 ; m / z 282\left(\mathrm{M}^{+}\right.$, $31 \%$ ), 241 (49), 240 (100), 239 (30), 212 (38), 195 (18), 148 (18), 121 (33), 120 (35), 119 (26), 93 (30) and 77 (20).
2-Isopropoxy-3-phenylpteridin-4(3H)-one 10d. Pale yellow solid ( $11.3 \mathrm{mg}, 36 \%$ ), mp $148-149{ }^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}, 282.1121$.
$\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{~N}_{4} \mathrm{O}_{2}$ requires $M, 282.1117$ ); $R_{\mathrm{F}} 0.30(\mathrm{~A}) ; v_{\text {max }}(\mathrm{K}-$ $\mathrm{Br}) / \mathrm{cm}^{-1}$ 2924, 1719, 1588, 1561, 1541, 1420, 1375, 1294, 1090, 916, 824 and $694 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.85(1 \mathrm{H}, \mathrm{d}, J 2.1), 8.70$ $(1 \mathrm{H}, \mathrm{d}, J 2.1), 7.49-7.60(3 \mathrm{H}, \mathrm{m}), 7.23-7.29(2 \mathrm{H}, \mathrm{m}), 5.62(1 \mathrm{H}$, sep, $J 6.2$ ) and $1.30(6 \mathrm{H}, \mathrm{d}, J 6.2) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 161.7$, $155.2,150.3,142.8,134.7,131.8,129.8,129.7,129.6,128.2$, 74.6 and $21.6 ; \mathrm{m} / \mathrm{z} 282\left(\mathrm{M}^{+}, 29 \%\right), 241$ (16), 240 (100), 239 (12), 212 (12), 195 (12), 148 (12), 121 (16), 120 (16), 119 (13) and 93 (13).
2-Diethylamino-3-phenylpteridin-4(3H)-one 10e. Pale yellow solid ( $67.8 \mathrm{mg}, 92 \%$ ), $\mathrm{mp} 154-155^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}, 295.1442$. $\mathrm{C}_{16} \mathrm{H}_{17} \mathrm{~N}_{5} \mathrm{O}$ requires $M, 295.1433$ ); $R_{\mathrm{F}} 0.23(\mathrm{~A}) ; v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1}$ 2990, 1701, 1555, 1528, 1478, 1415, 1385, 1279, 1213, 1078, 822 and $750 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.75(1 \mathrm{H}, \mathrm{d}, J 2.1), 8.54(1 \mathrm{H}, \mathrm{d}$, $J 2.1), 7.58-7.34(5 \mathrm{H}, \mathrm{m}), 3.28(4 \mathrm{H}, \mathrm{q}, J 7.1)$ and $0.91(6 \mathrm{H}, \mathrm{t}$, $J 7.1) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 162.8,156.7$, 155.3, 150.5, 141.5, $138.2,130.8,129.8,129.0,128.9,45.3$ and $12.4 ; m / z 295\left(\mathrm{M}^{+}\right.$, $34 \%$ ), 267 (17), 266 (100), 223 (10), 195 (14), 191 (9), 190 (23), 176 (18), 149 (14) and 119 (10).

2-Methoxy-3-propylpteridin-4(3H)-one 10f. Pale yellow solid $(26.7 \mathrm{mg}, 86 \%), \mathrm{mp} 123-124{ }^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}, 220.0968$. $\mathrm{C}_{10} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{O}_{2}$ requires $M, 220.0960$ ); $R_{\mathrm{F}} 0.22(\mathrm{~A}) ; v_{\text {max }}(\mathrm{KBr}) /$ $\mathrm{cm}^{-1} 2967,1703,1597,1559,1541,1453,1420,1265,1229,1192$, 988 and $739 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.83(1 \mathrm{H}, \mathrm{d}, J 2.2), 8.69(1$ $\mathrm{H}, \mathrm{d}, J 2.2), 4.24(3 \mathrm{H}, \mathrm{s}), 4.13(2 \mathrm{H}, \mathrm{t}, J 7.6), 1.76(2 \mathrm{H}$, sext, $J 7.6$, $1.8)$ and $1.00(3 \mathrm{H}, \mathrm{t}, J 7.4) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 161.5,156.4$, $154.2,150.1,142.9,131.2,56.8,44.2,21.4$ and $11.2 ; m / z$ $220\left(\mathrm{M}^{+}, 100 \%\right), 205(8), 192$ (9), 179 (26), 178 (34), 177 (19), 163 (13), 148 (13), 148 (62), 121 (9), 120 (48), 93 (12) and 79 (10).
In the case of 1-isopropyl isocyanate the major compound was 9 b not 10 g .
2-Isopropylamino-3-(1-methylethyl)pteridin-4(3H)-one 9b. Pale yellow solid ( $8.9 \mathrm{mg}, 35 \%$ ), mp $191-192^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}$, 247.1436. $\mathrm{C}_{12} \mathrm{H}_{17} \mathrm{~N}_{5} \mathrm{O}$ requires $M$, 247.1433); $R_{\mathrm{F}} 0.21$ (A); $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 3007,1711,1568,1535,1478,1420,1364,1269$, $1225,1182,1128,777$ and $737 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.67(1 \mathrm{H}$, d, $J 2.0), 8.46(1 \mathrm{H}, \mathrm{d}, J 2.0), 5.53(1 \mathrm{H}$, quint, $J 7.2), 4.80(1 \mathrm{H}, \mathrm{d}$, $J 6.8), 4.62(1 \mathrm{H}$, sep, $J 6.5), 1.60(6 \mathrm{H}, \mathrm{d}, J 7.2)$ and $1.36(6 \mathrm{H}, \mathrm{d}$, $J 6.5) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 162.3,155.6,151.5,150.3,140.6$, 129.7, 45.1, 44.4, 22.9 and 20.1; $m / z 247\left(\mathrm{M}^{+}, 100 \%\right)$, 205 (25), 204 (71), 190 (17), 164 (43), 163 (16), 162 (52), 149 (18), 147 (7), 136 (7), 120 (10), 119 (18) and 94 (9).
3-Benzyl-2-methoxypteridin-4(3H)-one 10h. White solid $(17.1 \mathrm{mg}, 44 \%), \mathrm{mp} 161-162{ }^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}, 268.0962$. $\mathrm{C}_{14} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{O}_{2}$ requires $M, 268.0960$ ); $R_{\mathrm{F}} 0.26(\mathrm{~A}) ; v_{\text {max }}(\mathrm{K} \mathrm{Br}) /$ $\mathrm{cm}^{-1}$ 3046, 2957, 1705, 1597, 1561, 1451, 1408, 1262, 1219, 1113, 988, 737, 710 and 694; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.82(1 \mathrm{H}, \mathrm{d}$, $J 2.0), 8.70(1 \mathrm{H}, \mathrm{d}, J 2.0), 7.41-7.47(2 \mathrm{H}, \mathrm{m}), 7.29-7.38(3 \mathrm{H}$, $\mathrm{m}), 5.34(2 \mathrm{H}, \mathrm{s})$ and $4.22(3 \mathrm{H}, \mathrm{s}) ; \delta_{\mathrm{c}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 161.7$, 156.2, 154.2, 150.3, 143.0, 135.9, 131.4, 129.047, 129.015, 128.5, 57.0 and $45.7 ; m / z 268\left(\mathrm{M}^{+}, 100 \%\right), 253$ (11), 236 (8), 225 (11), 208 (7), 163 (11), 148 (9), 120 (9), 104 (7) and 91 (34).
2-Methoxy-3-p-tolylpteridin-4(3H)-one 10i. White solid $(15.3 \mathrm{mg}, 55 \%), \mathrm{mp} 205-209^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}, 268.0971$. $\mathrm{C}_{14} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{O}_{2}$ requires $M, 268.0960$ ); $R_{\mathrm{F}} 0.28(\mathrm{~A}) ; v_{\max }(\mathrm{KBr}) /$ $\mathrm{cm}^{-1}$ 2957, 1711, 1603, 1449, 1408, 1356, 1294, 1092, 1040, 968,822 and $737 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.86(1 \mathrm{H}, \mathrm{d}, J 2.0), 8.72$ ( $1 \mathrm{H}, \mathrm{d}, J 2.0$ ), $7.32-7.38(2 \mathrm{H}, \mathrm{m}), 7.13-7.19(2 \mathrm{H}, \mathrm{m}), 4.11(3 \mathrm{H}$, s) and $2.45(3 \mathrm{H}, \mathrm{s}) ; \delta_{\mathrm{c}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 161.6,156.2,154.5$, $150.3,143.1,140.0,131.8,130.7,128.2,127.9,57.1$ and $21.4 ; m / z$ $268\left(\mathrm{M}^{+}, 100 \%\right), 267(28), 253$ (55), 251 (41), 237 (21), 209 (26), 133 (50), 132 (22), 105 (35), 104 (21) and 91 (25).
2-Methoxy-3-m-tolylpteridin-4(3H)-one 10j. Pale yellow solid $(26.4 \mathrm{mg}, 55 \%), \mathrm{mp}{ }^{191-192{ }^{\circ} \mathrm{C} \text { (Found: } \mathrm{M}^{+}, 268.0959 .}$ $\mathrm{C}_{14} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{O}_{2}$ requires: $M, 268.0960$ ); $R_{\mathrm{F}} 0.29$ (A); $v_{\text {max }}(\mathrm{K}-$ $\mathrm{Br}) / \mathrm{cm}^{-1} 2922,1717,1603,1561,1539,1443,1408,1350,1294$, $1194,1094,737$ and $694 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.86(1 \mathrm{H}, \mathrm{d}, J 2.1)$, $8.72(1 \mathrm{H}, \mathrm{d}, J 2.1), 7.40-7.48(1 \mathrm{H}, \mathrm{m}), 7.29-7.34(1 \mathrm{H}, \mathrm{m}), 7.05-$ $7.10(2 \mathrm{H}, \mathrm{m}), 4.11(3 \mathrm{H}, \mathrm{s})$ and $2.44(3 \mathrm{H}, \mathrm{s}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right)$
$161.5,156.1,154.5,150.3,143.1,140.2,134.3,131.8,130.7,129.8$, 128.7, 125.1, 57.1 and 21.4; $m / z 268\left(\mathrm{M}^{+}, 100 \%\right)$, 267 (12), 253 (22), 251 (21), 209 (11), 133 (18), 105 (9), 104 (8) and 91 (11).

2-Methoxy-3-o-tolylpteridin-4(3H)-one 10k. Pale yellow solid ( $8.3 \mathrm{mg}, 29 \%$ ), mp $144-146^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}, 268.0961$. $\mathrm{C}_{14} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{O}_{2}$ requires $M, 268.0960$ ); $R_{\mathrm{F}} 0.27$ (A); $v_{\text {max }}(\mathrm{K}-$ $\mathrm{Br}) / \mathrm{cm}^{-1} 2957,1713,1601,1562,1541,1447,1412,1354,1300$, 785 and $762 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.88(1 \mathrm{H}, \mathrm{d}, J 2.0), 8.74$ ( 1 $\mathrm{H}, \mathrm{d}, J 2.0), 7.33-7.44(3 \mathrm{H}, \mathrm{m}), 7.15-7.20(1 \mathrm{H}, \mathrm{m}), 4.11(3 \mathrm{H}, \mathrm{s})$ and $2.14(3 \mathrm{H}, \mathrm{s}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 161.0,156.1,154.8$, $150.4,143.2,135.8,133.7,131.8,131.7,130.2,128.3,127.7,57.2$ and $17.5 ; m / z 268\left(\mathrm{M}^{+}, 100 \%\right), 253(10), 252(16), 251(98), 238$ (10), 237 (67), 208 (16), 133 (14), 105 (10) and 104 (13).

2-Methoxy-3-(4-methoxyphenyl)pteridin-4(3H)-one 101. White solid ( $17.6 \mathrm{mg}, 55 \%$ ), mp $170-173^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}$, 284.0898. $\mathrm{C}_{14} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{O}_{3}$ requires $M, 284.0909$ ); $R_{\mathrm{F}} 0.25$ (A); $v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 2959,1711,1599,1561,1512,1445,1410,1354$, $1298,1250,1092$ and $1026 ; \delta_{H}\left(\mathrm{CDCl}_{3}, 500 \mathrm{MHz}\right) 8.86(1 \mathrm{H}, \mathrm{d}, J$ 2.0), $8.72(1 \mathrm{H}, \mathrm{d}, J 2.5), 7.17-7.20(2 \mathrm{H}, \mathrm{m}), 7.04-7.07(2 \mathrm{H}, \mathrm{m})$, $4.11(3 \mathrm{H}, \mathrm{s})$ and $3.88(3 \mathrm{H}, \mathrm{s}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 125 \mathrm{MHz}\right) 161.6$, $160.3,156.2,154.3,150.1,142.9,131.7,129.1,126.7,115.1,57.2$, $55.8 ; m / z 284\left(\mathrm{M}^{+}, 100 \%\right), 283$ (6), 269 (16), 253 (6), 241 (10), 226 (7), 225 (7), 149 (31), 134 (17), 121 (8) and 106 (8).

2-Methoxy-3-(3-methoxyphenyl)pteridin-4(3H)-one $\quad \mathbf{1 0 m}$. White solid ( $46.5 \mathrm{mg}, 81 \%$ ), mp $172-173^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}$, 284.0918. $\mathrm{C}_{14} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{O}_{3}$ requires $M$, 284.0909); $R_{\mathrm{F}} 0.26$ (A); $v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 3000,1719,1597,1561,1539,1447,1412,1354$, 1290, 1032 and 743 ; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.86(1 \mathrm{H}, \mathrm{d}, J 2.2)$, 8.72 ( $1 \mathrm{H}, \mathrm{d}, J 2.2$ ), $7.46(1 \mathrm{H}, \mathrm{t}, J 8.1)$, 7.05 ( 1 H , ddd, $J 8.4,2.6$, $0.9), 6.87(1 \mathrm{H}, \mathrm{ddd}, J 7.7,1.9,0.9), 6.82(1 \mathrm{H}, \mathrm{t}, J 2.0), 4.11(3 \mathrm{H}$, s) and $3.84(3 \mathrm{H}, \mathrm{s}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 161.4,160.9,156.0$, $154.5,150.3,143.1,135.4,131.8,130.7,120.3,115.4,114.1,57.1$ and 56.6; $m / z 284\left(\mathrm{M}^{+}, 100 \%\right), 283(16), 269(15), 267(13), 253$ (7), 226 (6), 225 (11), 149 (18) and 120 (6).

2-Methoxy-3-(2-methoxyphenyl)pteridin-4(3H)-one 10n. Pale yellow solid ( $30.3 \mathrm{mg}, 53 \%$ ) $\mathrm{mp} 166-168{ }^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}$, 284.0895. $\mathrm{C}_{14} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{O}_{3}$ requires $M, 284.0909$ ); $R_{\mathrm{F}} 0.25$ (A); $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 2928,1717,1605,1561,1539,1503,1445,1410$, $1354,1271,1119$, 1024 and $754 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.85(1 \mathrm{H}$, d, J2.1), $8.71(1 \mathrm{H}, \mathrm{d}, J 2.1), 7.45-7.53(1 \mathrm{H}, \mathrm{m}), 7.06-7.27(1 \mathrm{H}$, $\mathrm{m}), 4.10(3 \mathrm{H}, \mathrm{s})$ and $3.79,(3 \mathrm{H}, \mathrm{s}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 161.1$, $156.4,154.9,150.6,150.2,144.9,142.8,131.4,129.6,127.1$, 121.4, 112.5, 57.0 and $56.0 ; m / z 284\left(\mathrm{M}^{+}, 83 \%\right), 283$ (17), 269 (8), 255 (9), 254 (18), 253 (100), 225 (8), 149 (18), 134 (10), 120 (14) and 106 (11).

## 2-Methoxy-3-(3-trifluoromethylphenyl)pteridin-4(3H)-one

 100. Pale yellow solid ( $32.2 \mathrm{mg}, 48 \%$ ), $\mathrm{mp} 82-83{ }^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}, 322.0675 . \mathrm{C}_{14} \mathrm{H}_{9} \mathrm{~F}_{3} \mathrm{~N}_{4} \mathrm{O}_{2}$ requires $M, 322.0678$ ); $R_{\mathrm{F}} 0.07$ (H:A $1: 1$ ); $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 1723,1607,1562,1541,1447,1412$, 1352, 1333, 1292, 1177, 1138 and 1098; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right)$ $8.89(1 \mathrm{H}, \mathrm{d}, J 2.0), 8.74(1 \mathrm{H}, \mathrm{d}, J 2.0), 7.66-7.81(2 \mathrm{H}, \mathrm{m}), 7.49-$ $7.59(2 \mathrm{H}, \mathrm{m})$ and $4.13(3 \mathrm{H}, \mathrm{s})$; $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 161.3$, $155.5,154.5,150.6,143.4,134.9,132.6$ (q, J33.2), 132.0, 131.6, 130.7, $126.8(\mathrm{q}, J 3.5), 125.6(\mathrm{q}, J 3.9), 123.7(\mathrm{q}, J 271.7)$ and 57.3; $m / z 322\left(\mathrm{M}^{+}, 100 \%\right), 307$ (14), 305 (29), 277 (9), 264 (8), 263 (15), 187 (27), 159 (16), 149 (8), 145 (8) and 120 (8).3-(4-Chlorophenyl)-2-methoxypteridin-4(3H)-one 10p. Pale yellow solid ( $46.7 \mathrm{mg}, 77 \%$ ) mp 153- $154^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}$, 288.0415. $\mathrm{C}_{13} \mathrm{H}_{9} \mathrm{ClN}_{4} \mathrm{O}_{2}$ requires $M, 288.0414$ ); $R_{\mathrm{F}} 0.13$ ( $\mathrm{H}: \mathrm{A}$ $1: 2) ; v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 1715,1605,1562,1541,1493,1476,1446$, $1412,1354,1296$ and $1090 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.88(1 \mathrm{H}$, br), $8.73(1 \mathrm{H}, \mathrm{br}), 7.53(2 \mathrm{H}, \mathrm{d}, J 8.8), 7.24(2 \mathrm{H}, \mathrm{d}, J 8.8)$ and 4.12 $(3 \mathrm{H}, \mathrm{s}) ; \delta_{\mathrm{c}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 161.4,155.9,154.4,150.5,143.2$, $135.9,132.8,131.6,130.3,129.7$ and $57.2 ; m / z 290(M+2$, $32 \%$ ), $289(\mathrm{M}+1,17), 288\left(\mathrm{M}^{+}, 100\right), 273$ (18), 271 (15), 257 (7), 229 (9), 155 (8), 153 (25), 148 (7) and 125 (19).

2-Methoxy-3-(1-naphthyl)pteridin-4(3H)-one 10q. White solid $\left(9.3 \mathrm{mg}, 30 \%\right.$ ), mp $154-155^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}, 304.0968$. $\mathrm{C}_{17} \mathrm{H}_{12} \mathrm{~N}_{4} \mathrm{O}_{2}$ requires $M, 322.0960$ ); $R_{\mathrm{F}} 0.26(\mathrm{~A}) ; v_{\text {max }}(\mathrm{K}-$ $\mathrm{Br}) / \mathrm{cm}^{-1} 1719,1655,1603,1561,1541,1410,1348,1117$ and
$775 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.92(1 \mathrm{H}, \mathrm{d}, J 2.1), 8.77(1 \mathrm{H}, \mathrm{d}, J$ 2.1), 7.96-8.06 ( $2 \mathrm{H}, \mathrm{m}$ ), 7.44-7.68 ( $5 \mathrm{H}, \mathrm{m}$ ) and $4.03(3 \mathrm{H}, \mathrm{s})$; $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 156.6,150.5,143.3,134.9,131.9,131.2$, $130.6,129.8,129.2,128.0,127.1,126.6,126.0,121.7$ and 57.2 (two quaternary carbons were not detected); $m / z 304$ ( $\mathrm{M}^{+}$, $100 \%$ ), 289 (3), 287 (10), 273 (5), 245 (8), 169 (22), 141 ( 8 ), 140 (8), 127 (3) and 114 (3).

2-Isopropylamino-3-phenylpteridin-4(3H)-one 12a. Pale yellow solid ( $58.9 \mathrm{mg}, 82 \%$ ), $\mathrm{mp} 222-224^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}$, 281.1278. $\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{O}$ requires $M, 281.1277$ ); $R_{\mathrm{F}} 0.20$ (A); $v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1} 2975,2928,1703,1578,1561,1530,1474,1414$, $1281,1210,1127,733$ and $696 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.73(1 \mathrm{H}$, d, $J 2.2), 8.50(1 \mathrm{H}, \mathrm{d}, J 2.0), 7.60-7.72(3 \mathrm{H}, \mathrm{m}), 7.32-7.37(2 \mathrm{H}$, $\mathrm{m}), 4.47(1 \mathrm{H}$, sepd, $J 6.6$ and 1.2), $4.23(1 \mathrm{H}, \mathrm{dd}, J 6.6$ and 1.2$)$ and $1.45(6 \mathrm{H}, \mathrm{d}, J 6.6) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 161.6,156.4,151.5$, $150.4,140.6,134.0,131.4,130.9,129.4,128.9,44.3$ and 22.6; $m / z 281\left(\mathrm{M}^{+}, 47 \%\right), 280(4), 266(5), 239(20), 238(100), 195(7)$, 169 (4), 119 (13), 118 (21) and 77 (15).
2-Allylamino-3-phenylpteridin-4(3H)-one 12b. Pale yellow solid ( $24.9 \mathrm{mg}, 45 \%$ ), $\mathrm{mp} 217-218{ }^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}, 279.1125$. $\mathrm{C}_{15} \mathrm{H}_{13} \mathrm{~N}_{5} \mathrm{O}$ requires $M, 279.1120$ ); $R_{\mathrm{F}} 0.10(\mathrm{~A}) ; v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1}$ 2922, 1701, 1582, 1562, 1532, 1472, 1414, 1287, 1209, 1101, 731 and 698; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.74(1 \mathrm{H}, \mathrm{d}, J 2.0), 8.52(1 \mathrm{H}, \mathrm{d}$, $J 2.0), 7.59-7.71(3 \mathrm{H}, \mathrm{m}), 7.35-7.40(2 \mathrm{H}, \mathrm{m}), 5.88(1 \mathrm{H}, \mathrm{ddt}$, $J 16.9,10.5$ and 5.4$), 5.13(1 \mathrm{H}, \mathrm{dq}, J 10.8$ and 1.3), $5.10(1 \mathrm{H}$, $\mathrm{dq}, J 16.9$ and 1.5$), 4.58(1 \mathrm{H}, \mathrm{t}, J 5.4)$ and $4.20(2 \mathrm{H}, \mathrm{tt}, J 5.6$ and 1.6); $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 161.5,156.3,152.1,150.4,141.0$, 133.9, 133.7, 131.5, 131.0, 130.4, 129.0, 117.1 and 44.4; $m / z 279$ $\left(\mathrm{M}^{+}, 100 \%\right), 278(40), 265(10), 264(60), 238(8), 195(6), 169$ (6), 119 (7), 118 (6) and 77 (17).

3-Allyl-2-anilinopteridin-4(3H)-one 13b. Pale yellow solid $(21.6 \mathrm{mg}, 39 \%), \mathrm{mp} 112-113^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}, 279.1120$. $\mathrm{C}_{15} \mathrm{H}_{13} \mathrm{~N}_{5} \mathrm{O}$ requires $M, 279.1120$ ); $R_{\mathrm{F}} 0.20(\mathrm{~A}) ; v_{\text {max }}(\mathrm{KBr}) / \mathrm{cm}^{-1}$ 2926, 1686, 1561, 1528, 1451, 1412, 1223, 1074 and 691; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.77(1 \mathrm{H}, \mathrm{d}, J 2.0), 8.58(1 \mathrm{H}, \mathrm{d}, J 2.0)$, 7.58-7.63 ( $2 \mathrm{H}, \mathrm{m}$ ), 7.37-7.46 ( $2 \mathrm{H}, \mathrm{m}$ ), 7.16-7.24 ( $1 \mathrm{H}, \mathrm{m}$ ), 7.00 $(1 \mathrm{H}, \mathrm{br}), 6.01(1 \mathrm{H}, \mathrm{ddt}, J 17.4,10.0$ and 5.4$)$, $5.58(1 \mathrm{H}, \mathrm{dq}, J$ 10.0 and 0.8 ), $5.56(1 \mathrm{H}, \mathrm{dq}, J 17.4$ and 0.8$)$ and $4.01(2 \mathrm{H}, \mathrm{dt}, J$ 5.6 and 1.7); $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 161.6,156.9,155.3,150.7$, $150.5,141.9,137.3,132.2,129.7,125.7,122.4,120.0$ and 45.1 ; $m / z 279\left(\mathrm{M}^{+}, 100 \%\right), 278$ (89), 265 (9), 264 (55), 251 (10), 250 (40), 238 (16), 236 (11), 224 (30), 195 (9), 188 (14), 117 (12) and 77 (32).
3-Phenyl-2-prop-2-ynylaminopteridin-4(3H)-one 12c. Yellow solid ( $18.1 \mathrm{mg}, 32 \%$ ), $\mathrm{mp} 212-215^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}, 277.0964$. $\mathrm{C}_{15} \mathrm{H}_{11} \mathrm{~N}_{4} \mathrm{O}_{2}$ requires $\left.M, 277.0964\right)$; $R_{\mathrm{F}} 0.17(\mathrm{~A}) ; v_{\text {max }}(\mathrm{K}-$ $\mathrm{Br}) / \mathrm{cm}^{-1} 3057,2924,1701,1615,1584,1562,1534,1472,1289$, $1256,1211,750$ and $669 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.77(1 \mathrm{H}, \mathrm{d}, J$ 2.2), $8.56(1 \mathrm{H}, \mathrm{d}, J 2.2), 7.61-7.72(3 \mathrm{H}, \mathrm{m}), 7.35-7.40(2 \mathrm{H}, \mathrm{m})$, $4.68(1 \mathrm{H}, \mathrm{t}, J 5.2), 4.37(2 \mathrm{H}, \mathrm{dd}, J 5.2$ and 2.6$)$ and $2.23(1 \mathrm{H}, \mathrm{t}$, $J$ 2.6); $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 161.3,155.9,151.6,150.5,141.4$, 133.6, 131.5, 131.1, 130.5, 129.0, 79.0, 72.7 and 32.1; m/z 277 ( $\mathrm{M}^{+}, 24 \%$ ), 276 (18), 250 (10), 249 (66), 248 (100), 208 (10), 131 (11), 118 (9), 104 (7), 91 (6) and 77(17).

2-Allylamino-3-(4-chlorophenyl)pteridin-4(3H)-one 12d. Yellow solid ( $70.4 \mathrm{mg}, 55 \%$ ), mp 233-235 ${ }^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}$, $313.0715 . \mathrm{C}_{15} \mathrm{H}_{12} \mathrm{ClN}_{5} \mathrm{O}$ requires $M, 313.0730$ ); $R_{\mathrm{F}} 0.17(\mathrm{H}: \mathrm{A}$ $1: 2) ; v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 3021,1705,1588,1562,1537,1491,1472$, $1412,1208,1094$ and $928 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.74(1 \mathrm{H}, \mathrm{d}, J$ 2.1 ), 8.52 ( $1 \mathrm{H}, \mathrm{d}, J 2.1$ ), 7.59-7.66 ( $2 \mathrm{H}, \mathrm{m}$ ), 7.29-7.36 ( $2 \mathrm{H}, \mathrm{m}$ ), $5.89(1 \mathrm{H}$, ddt, $J 16.8,10.6$ and 5.6$), 5.15(1 \mathrm{H}, \mathrm{dq}, J 10.8$ and $1.3), 5.13(1 \mathrm{H}, \mathrm{dq}, J 16.8$ and 1.3$), 4.56(1 \mathrm{H}, \mathrm{t}, J 5.6)$ and $4.20(2$ $\mathrm{H}, \mathrm{tt}, J 5.6$ and 1.5); $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 161.4,156.2,151.8$, $150.6,141.1,137.3,133.5,132.3,131.7,130.4,130.2,117.4$ and 44.5; m/z $315(\mathrm{M}+2,32 \%), 314(\mathrm{M}+1,31), 313\left(\mathrm{M}^{+}, 100\right)$, 312 (42), 300 (24), 299 (13), 298 (76), 229 (6), 203 (6), 188 (6), 152 (8), 151 (6), 119 (7), 111 (10) and 75 (6).

3-Allyl-2-(4-chlorophenylamino)pteridin-4(3H)-one 13d. Pale yellow solid ( $20.5 \mathrm{mg}, 16 \%$ ), $\mathrm{mp} 87-88^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}$, 313.0732. $\mathrm{C}_{15} \mathrm{H}_{12} \mathrm{ClN}_{5} \mathrm{O}$ requires $M, 313.0730$ ); $R_{\mathrm{F}} 0.24$ (H:A
$1: 2) ; v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 2924,1686,1603,1561,1528,1493,1460$, 1422 and $824 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.78(1 \mathrm{H}, \mathrm{d}, J 2.0), 8.60(1$ $\mathrm{H}, \mathrm{d}, J 2.0), 7.50-7.60(2 \mathrm{H}, \mathrm{m}), 7.33-7.41(2 \mathrm{H}, \mathrm{m}), 6.98(1 \mathrm{H}$, br), $6.10(1 \mathrm{H}$, ddt, $J 17.1,10.7$ and 5.4$), 5.58(1 \mathrm{H}, \mathrm{dq}, J 10.8$ and 0.7$), 5.56(1 \mathrm{H}, \mathrm{dq}, J 17.1$ and 1.0$)$ and $5.00(2 \mathrm{H}, \mathrm{dt}, J 5.4$ and 1.7); $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 161.5,155.1,150.7,150.4,142.1$, $135.9,132.2,131.0,130.3,129.7,123.7,120.0$ and $45.1 ; \mathrm{m} / z 315$ $(\mathrm{M}+2,32 \%), 314(\mathrm{M}+1,34), 313\left(\mathrm{M}^{+}, 100\right), 312(62), 300$ (19), 298 (64), 284 (32), 276 (27), 274 (94), 258 (23), 188 (24), 153 (18), 121 (19), 111 (16), 93 (35) and 66 (16).

3-(4-Chlorophenyl)-2-prop-2-ynylaminopteridin-4(3H)-one 12e. Yellow solid ( $16.0 \mathrm{mg}, 18 \%$ ), mp 131-132 ${ }^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}$ 311.0562. $\mathrm{C}_{15} \mathrm{H}_{10} \mathrm{ClN}_{5} \mathrm{O}$ requires $M, 311.0574$ ); $R_{\mathrm{F}} 0.17$ (A); $v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 2926,1701,1588,1561,1530,1474,1412,1263$, 1090 and $1017 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.76(1 \mathrm{H}, \mathrm{d}, J 2.1), 8.55(1$ H, d, J2.1), 7.59-7.64 ( $2 \mathrm{H}, \mathrm{m}$ ), 7.29-7.36 ( $2 \mathrm{H}, \mathrm{m}$ ), $4.75(1 \mathrm{H}, \mathrm{t}$, $J 5.2), 4.37(2 \mathrm{H}, \mathrm{dd}, J 5.2$ and 2.5$)$ and $2.25(1 \mathrm{H}, \mathrm{t}, J 2.5)$; $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 161.3,155.9,151.3,150.6,141.5,137.4$, $132.0,131.7,130.4,130.3,78.9,72.8$ and $32.1 ; m / z 313(\mathrm{M}+2$, $8 \%$ ), $312(\mathrm{M}+1,13), 311\left(\mathrm{M}^{+}, 25\right), 310(26), 285(18), 284$ (38), 283 (55), 282 (100), 248 (8), 242 (11), 152 (12), 131 (15) and 111 (9).

2-Allylamino-3-(4-methoxyphenyl)pteridin-4(3H)-one 12 f . Yellow solid ( $28.8 \mathrm{mg}, 28 \%$ ), mp 153-154 ${ }^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}$, 309.1224. $\mathrm{C}_{16} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{O}_{2}$ requires $M, 309.1226$ ); $R_{\mathrm{F}} 0.09$ (A); $v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 2926,1701,1584,1562,1530,1510,1470,1414$, $1252,1107,1028$ and $824 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.72(1 \mathrm{H}, \mathrm{d}, J$ $2.0), 8.50(1 \mathrm{H}, \mathrm{d}, J 2.0), 7.24-7.31(2 \mathrm{H}, \mathrm{m}), 7.08-7.15(2 \mathrm{H}, \mathrm{m})$, $5.89(1 \mathrm{H}$, ddt, $J 16.7,10.9$ and 5.5$)$, $5.13(1 \mathrm{H}, \mathrm{dq}, J 10.9$ and $1.3), 5.12(1 \mathrm{H}, \mathrm{dq}, J 16.7$ and 1.5$), 4.71(2 \mathrm{H}, \mathrm{t}, J 5.5), 4.20(2 \mathrm{H}$, $\mathrm{tt}, J 5.6$ and 1.6 ) and $3.88(3 \mathrm{H}, \mathrm{s}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 161.8$, $161.3,156.2,152.2,150.3,140.7,133.7,130.3,130.0,125.9$, 117.0, 116.5, 55.7 and 44.3; m/z $309\left(\mathrm{M}^{+}, 100 \%\right)$, 308 (38), 295 (13), 294 ( 81 ), 253 (6), 202 (6), 148 (11), 147 (13), 146 (6), 133 (13), 108 (13) and 92 (6).

3-Allyl-2-(4-methoxyphenylamino)pteridin-4(3H)-one 13f. Yellow solid ( $20.0 \mathrm{mg}, 20 \%$ ) $\mathrm{mp} 86-88{ }^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}$, 309.1231. $\mathrm{C}_{16} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{O}$ requires $M, 309.1226$ ); $R_{\mathrm{F}} 0.16$ (A); $v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 2926,1690,1613,1562,1528,1510,1460,1424$, $1244,1107,1024$ and $826 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.72(1 \mathrm{H}, \mathrm{d}, J$ $2.0), 8.53(1 \mathrm{H}, \mathrm{d}, J 2.0), 7.42-7.47(2 \mathrm{H}, \mathrm{m}), 6.98-6.94(2 \mathrm{H}, \mathrm{m})$, $6.94(1 \mathrm{H}, \mathrm{s}), 6.02(1 \mathrm{H}, \mathrm{ddt}, J 17.6,10.0$ and 5.4$), 5.508(1 \mathrm{H}, \mathrm{d}, J$ 17.2), $5.504(1 \mathrm{H}, \mathrm{d}, J 10.8), 4.98(2 \mathrm{H}, \mathrm{dt}, J 5.4$ and 1.6 ) and 3.81 $(3 \mathrm{H}, \mathrm{s}) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 157.9,155.5,152.1,151.0,150.6$, $141.5,132.0,130.0,127.1,124.8,120.0,114.8,55.7$ and $44.9 ; m / z$ $309\left(\mathrm{M}^{+}, 100 \%\right.$ ), 308 (29), 295 (12), 294 (65), 293 (9), 281 (27), 280 (20), 266 (17), 254 (21), 148 (11), 147 (20), 146 (10) and 108 (9).

## Synthesis of imidazo[1,2-a] pteridine derivatives 15 by iodoimidazolination of 2-allylaminopteridin-4(3H)-one derivatives 12: general procedure

To a solution of the 2-allylaminopteridin-4(3H)-one 12a (10.6 $\mathrm{mg}, 0.038 \mathrm{mmol}$ ) in THF ( $5 \mathrm{~cm}^{3}$ ) was added iodine ( 20 mg , $0.079 \mathrm{mmol}, 2.1$ equiv.) and sodium hydrogen carbonate ( 7.4 $\mathrm{mg}, 0.088 \mathrm{mmol}, 2.3$ equiv.). The mixture was stirred at room temperature under nitrogen until starting material had disappeared (TLC). The reaction mixture was treated with saturated aqueous sodium sulfite to reduce the excess of iodine after which it was diluted with water and extracted with AcOEt $\left(20 \mathrm{~cm}^{3} \times 3\right)$. The combined layer extracts were dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under reduced pressure and the residue was purified on a silica gel column ( $\mathrm{H}: \mathrm{A} 1: 2, \mathrm{v} / \mathrm{v}$, as eluent) to afford the imidazo $[1,2-a]$ pteridine derivative 15 a ( 15.2 mg , $0.038 \mathrm{mmol}, 99 \%$ ).

1-Iodomethyl-4-phenyl-1,2-dihydroimidazo[1,2-a] pteridin$\mathbf{5 ( 4 H})$-one 15a. Yellow solid ( $15.2 \mathrm{mg}, 99 \%$ ), mp 122-123 ${ }^{\circ} \mathrm{C} ; R_{\mathrm{F}}$ $0.15(\mathrm{~A}) ; v_{\text {max }}\left(\mathrm{CHCl}_{3}\right) / \mathrm{cm}^{-1} 1717,1649,1545,1493,1422,1074$, 930 and $791 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200 \mathrm{MHz}\right) 8.42(2 \mathrm{H}, \mathrm{s}), 7.47-7.60(3 \mathrm{H}$, m), $7.35-7.39(2 \mathrm{H}, \mathrm{m}), 4.70-4.82(1 \mathrm{H}, \mathrm{m}), 4.17(1 \mathrm{H}, \mathrm{dd}, J 14.8$
and 9.7), $3.93(1 \mathrm{H}, \mathrm{dd}, J 10.2$ and 6.4$), 3.74(1 \mathrm{H}, \mathrm{dd}, J 14.8$ and $5.0)$ and $3.61(1 \mathrm{H}, \mathrm{dd}, J 10.2$ and 2.4$) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right)$ $159.5,150.9,148.2,147.8,139.2,135.4,130.2,129.8,128.7$, 128.5, 59.0, 58.2 and $9.1 ; m / z 405\left(\mathrm{M}^{+}, 100 \%\right), 404$ (45), 279 (13), 278 (77), 277 (25), 276 (22), 265 (5), 264 (36), 132 (5), 131 (15) and 77 (22). In the HRMS, the $M^{+}$peak of 15a (405.0087) was hidden by one of the perfluorocarbon peaks ( $\mathrm{C}_{10} \mathrm{~F}_{15}$, 404.9760) used as a standard.

4-(4-Chlorophenyl)-1-iodomethyl-1,2-dihydroimidazo [1,2-a]-pteridin-5(4H)-one 15b. Yellow solid ( $36.4 \mathrm{mg}, 100 \%$ ), mp $79-82^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}$, 438.9697. $\mathrm{C}_{15} \mathrm{H}_{11} \mathrm{ClIN}_{5} \mathrm{O}$ requires: $M$, 438.9695); $R_{\mathrm{F}} 0.36(\mathrm{~A}) ; v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 1711,1647,1574,1541$, 1493, 1454, 1420, 1196, 1090,858 and $754 ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 200\right.$ $\mathrm{MHz}) 8.43(2 \mathrm{H}, \mathrm{s}), 7.48-7.55(2 \mathrm{H}, \mathrm{m}), 7.28-7.35(2 \mathrm{H}, \mathrm{m})$, 4.49-4.81 ( $1 \mathrm{H}, \mathrm{m}$ ), $4.16(1 \mathrm{H}, \mathrm{dd}, J 14.7$ and 9.7$), 3.95(1 \mathrm{H}$, dd, $J 10.3$ and 6.1), $3.73(1 \mathrm{H}, \mathrm{dd}, J 14.7$ and 5.1$)$ and $3.59(1 \mathrm{H}, \mathrm{dd}, J$ 10.3 and 2.3); $\delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 159.4,150.6,148.1,147.9$, 139.2, 135.8, 133.8, 130.4, 129.9, 128.5, 58.9, 58.1 and $9.2 ; \mathrm{m} / \mathrm{z}$ $441(\mathrm{M}+2,32 \%), 440(\mathrm{M}+1,24), 439\left(\mathrm{M}^{+}, 100\right), 438(22)$, 314 (26), 313 (30), 312 (89), 311 (48), 310 (28), 300 (12), 298 (39), 165 (13) and 111 (11).

1-Iodomethyl-4-(4-methoxyphenyl)-1,2-dihydroimidazo[1,2-a]-pteridin-5(4H)-one 15c. Yellow solid ( $28.6 \mathrm{mg}, 87 \%$ ), mp $83-85{ }^{\circ} \mathrm{C}$ (Found: $\mathrm{M}^{+}, 435.0183 . \mathrm{C}_{16} \mathrm{H}_{14} \mathrm{IN}_{5} \mathrm{O}_{2}$ requires $M$, 435.0192); $R_{\mathrm{F}} 0.13(\mathrm{~A}) ; v_{\max }(\mathrm{KBr}) / \mathrm{cm}^{-1} 2928,1711,1645,1543$, 1512, 1493, 1454, 1420, 1300, 1250, 1196, 1030 and 750 ; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right) 8.42(2 \mathrm{H}, \mathrm{s}), 7.24-7.32(2 \mathrm{H}, \mathrm{m}), 7.00-7.08$ $(2 \mathrm{H}, \mathrm{m}), 4.69-4.81(1 \mathrm{H}, \mathrm{m}), 4.16(1 \mathrm{H}, \mathrm{dd}, J 14.8$ and 9.8$), 3.92$ $(1 \mathrm{H}, \mathrm{dd}, J 10.2$ and 6.4$), 3.81(3 \mathrm{H}, \mathrm{s}), 3.73(1 \mathrm{H}, \mathrm{dd}, J 14.8$ and $5.1)$ and $3.61(1 \mathrm{H}, \mathrm{dd}, J 10.2$ and 2.4$) ; \delta_{\mathrm{C}}\left(\mathrm{CDCl}_{3}, 50 \mathrm{MHz}\right)$ $160.4,159.7,151.4,151.1,148.1,147.7,139.1,129.4,127.9$, 115.4, 59.0, 58.2, 55.6 and $9.1 ; m / z 435\left(\mathrm{M}^{+}, 87 \%\right), 434$ (19), 309 (16), 308 (75), 307 (100), 306 (73), 294 (23), 292 (8), 279 (9), 278 (13), 264 (13), 161 (8) and 77 (8).

## X-Ray crystal structure analysis of $\mathbf{6}^{29}$

$\mathrm{C}_{24} \mathrm{H}_{20} \mathrm{~N}_{3} \mathrm{O}_{2} \mathrm{P}, \quad M=413.41$, Orthorhombic, space group $P 2_{1} 2_{1} 2_{1}, \quad a=14.76(3), \quad b=32.69(2), \quad c=9.06(1) ~ \AA \AA, \quad V=$ $4373(11) \AA^{3}, Z=8.0, D_{\mathrm{c}}=1.256 \mathrm{~g} \mathrm{~cm}^{-3}$. A white, hygroscopic prism $(0.22 \times 0.48 \times 0.66 \mathrm{~mm})$ was mounted on a Rigaku AFC5S diffractometer with graphite-monochromated Mo-K $\alpha$ radiation ( $\lambda=0.71069 \AA$ ). Data collection using the $\omega$ scan technique to a maximum $2 \theta$ value of $54.9^{\circ}$ gave 5581 reflections. The structure was solved by direct method and refined by fullmatrix least squares technique (TEXSAN ${ }^{30}$ system as the computer program, and MITHRIL ${ }^{31}$ as the structure solution method). The non-hydrogen atoms were refined anisotropically. The hydrogen atoms were included by calculation and were not refined. Final cycle of the refinement was calculated by 1384 observed reflections $[I>3.00 \sigma(I)]$ and 541 variable parameters and $R$ and $R_{\mathrm{w}}$ values are 0.063 and 0.065 . Atomic coordinates, bond lengths and angles, and thermal parameters have been deposited with the Cambridge Crystallographic Data Centre. ${ }^{29}$

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